WINTER - 2014

PURUS - IV (2011 COURSE): SUBJECT: PHARMACEUTICAL ANALYSIS - II

Time: 2:00 P.M.To 5:00 Day: Thussolay
Date: 13-11-2014 Max Marks. 80 N.B. 1) Q. No. 1 & Q. No. 5 are COMPULSORY. Out of the remaining solve any TWO questions from Section - I and Section - II each. 2) Answers to the two sections should be written in SEPARATE answer books. 3) Figures to the right indicate FULL marks. **SECTION-I** (10)Attempt Any FIVE of the following 01. Types of buffers and composition used for calibration of potentiometer. a) State and explain terms in Ilkovic equation. b) Define Linearly polarized light and Elliptically polarized light. c) Organic applications of polarography. d) Enlist drugs assayed by potentiometric titrations. e) Write general components in Instrumental analysis. f) Name different parts of polarimeter. State functions of each. Discuss (08) 02 applications of polarimetry in detail. Explain methods to detect end point potentiometrically. Discuss (07) b) potentiometric titrations. Q3 Explain an ideal polarogram. Write in detail factors affecting diffusion (08) current. Explain polarographic apparatus with its working. (07)b) Write short notes on Any THREE Q4 (15)Classification and advantages of instrumental methods of analysis. a) b) Glass electrode.

Applications of Polarography.

Saccharimeter.

c)

SECTION - II

	Attempt Any FIVE of the following	(10)
a)	Define equivalent conductance and molar conductance.	
b)	Define solubility product and enlist factors affecting it.	
c)	Explain specific and molar refraction.	
d)	Write advantages of rotating platinum electrode over dropping mercury	
	electrode.	
e)	Classify amperometric titrations.	
f)	Define fractional precipitation and co-precipitation.	
a)		(08)
b)	Explain principle of gravimetric analysis and add a note on apparatus needed	(07)
	for amperometric titration.	
		(0.0)
,		(08)
b)		(07)
	refractometers.	
	Write short notes on ANY THREE	(15)
a)	Applications of conductometry.	
b)	Amperometric titrations.	
c)	Applications of gravimetric analysis in Pharmaceuticals.	
d)	Applications of refractometry.	
	b) c) d) e) f) a) b) b) c) c)	 a) Define equivalent conductance and molar conductance. b) Define solubility product and enlist factors affecting it. c) Explain specific and molar refraction. d) Write advantages of rotating platinum electrode over dropping mercury electrode. e) Classify amperometric titrations. f) Define fractional precipitation and co-precipitation. a) Discuss conductometric titration curves in detail. b) Explain principle of gravimetric analysis and add a note on apparatus needed for amperometric titration. a) Explain unit operations in gravimetric analysis. b) Classify types of refractometers. Explain instrumentation of any two refractometers. Write short notes on ANY THREE a) Applications of conductometry. b) Amperometric titrations. c) Applications of gravimetric analysis in Pharmaceuticals.

WINTER - 2014 PURUS - IV (2011 COURSE): . SUBJECT: PHARMACEUTICAL MICROBIOLOGY (INCLUDING IMMUNOLOGY) - II Time: 2:00 P.M: To 5:00 P.M. Max. Marks: 80 :Thursday Date :20-11-2014 N.B.: Q.No.1 and Q.No.5 are COMPULSORY. Out of remaining questions attempt ANY TWO questions from each section. Answers to both the sections should be written in the SEPARATE answer books. 2) Draw neat and labeled diagrams WHEREVER necessary. 3) Figures to the right indicate FULL marks. **SECTION-I** [10] Q.1 Attempt ANY FIVE of the following: a) Enlist sources of microbial spoilage. b) Define MIC. How it is determined? c) How screening of microbial species is performed? d) Which tests detect Salmonella contamination? e) Give significance of microbial assays. Draw a well labeled diagram of Tray Fermenter. 0.2 Attempt the following: [15] a) Discuss Preservative Efficacy Tests. b) Correlate MIC and Microbial Assays. Discuss Microbial Assay of Vitamins. 0.3 a) Draw and discuss Fluidized Bed Bioreactor. [08] b) How penicillin is produced by fermentation? [07] Write short notes on ANY THREE of the following: Q.4 [15] a) Microbial Assay of Streptomycin b) Probiotics c) Microbial Limit Tests d) Air-Lift Fermenters **SECTION - II** Attempt ANY FIVE of the following: [10] a) Compare immediate and delayed hypesensitivity. b) Define and explain Adjuvants. c) Classify immunological products. d) Enlist Type I hypersensitivity reactions. What is Humoral Immune Response? How is 'Virulence' determined? Attempt the following: [15] Q.6 a) Discuss Quality Control of vaccines. b) Explain Radioimmunoassay in detail. c) Describe the risk of Rh-incompatibility in newborns. Q.7 a) Give an exhaustive account of precipitation reactions. [08] b) Write in detail about Immunoglobulins. [07]

Write short notes on ANY THREE of the following:

[15]

Q.8

a) Phagocytosis

b) Monoclonal Antibodies

WINTER - 2014

PURUS -IV (2011 COURSE): SUBJECT: PHYSICAL PHARMACY-II

Day: Saturday

Viscoelasticity.

Measurement of diffraction angle.

b)

Time: 2:00 P. M : To 5:00

Max. Marks: 80 Date: 15-11-2014 N.B.: Q.No.1 and Q.No.5 are COMPULSORY. Out of remaining attempt ANY TWO 1) questions from each Section. Answer to both the sections should be written in SEPARATE answer books. 2) Figure to the right indicates FULL marks. 3) SECTION-I [10] 0.1 Answer ANY FIVE of the following: What is micellar solubilization? a) Differentiate between flocculated and deflocculated suspension. b) Write name and formula for equation governing settling of particles. What is significance of zeta potential? Explain the effect of electrolytes on zeta potential. Write formula for Freundlich adsorption isotherm. e) What are sedimentation parameters? Classify different types of colloids and discuss their distinguishing features [07] 0.2 a) and properties. [08] What are surfactants? Explain in detail their applications in pharmacy. b) Explain in detail factors influencing settling of particles in pharmaceutical [07] Q.3 a) Derive an equation for spreading co - efficient. Give its significance. [08] [15] Q.4 Answer ANY THREE of the following: Methods for measurement of surface tension. Electric Double layer. Physical stability of emulsions. Kinetic properties of colloids. **SECTION-II** [10] 0.5 Answer ANY FIVE of the following: What is crystal habit? a) Define: i) Angle of repose ii) Carr's consolidation index b) Classify various interparticulate bonds observed in a tablet. c) Enlist methods used for measurement of particle surface area. Explain dilatant flow behaviour. Explain different types of packing arrangements in powder. 0.6 Define Thixotropy and explain the mechanism of the same. Add a brief note on mechanisms of compression of particles and methods to [08] evaluate the same. Write a note different types of viscometers used to measure viscosity of 0.7 a) Explain in detail methods used to determine particle size. [08] b) Answer ANY THREE of the following: [15] Q.8 Phenomenon of compression.

PURUS - IV (2011 COURSE): WINTER - 2014 SUBJECT: PHARMACOLOGY - I Time: 2:00 P.M. TO 5-00 P.M :Saturday Day Max. Marks: 80 Date : 22-11-2014 N.B.: Q.No.1 and Q.No.5 are COMPULSORY. Out of remaining questions attempt 1) ANY TWO questions from each section. Answers to both the sections should be written in the SEPARATE answer books. 2) 3) Figures to the right indicate FULL marks. **SECTION-I** Attempt ANY FIVE of the following: [10] Pharmacotherapeutics b) Placebo Spansules Bioequivalence Clearance Volume of distribution Enlist the routes of drug administration. Explain oral route in detail. Add a note [08] on first pass effect. Explain the phases of drug metabolism. What are the factors affecting drug [07] metabolism? Define drug-drug interaction. Explain pharmacodynamic drug interaction in [08] b) Explain the factors which affect drug action. [07] Write short notes on ANY THREE of the following: [15] Nature of drugs Blood Brain Barrier b) Therapeutic Window and its importance c) Tolerance SECTION - II Q.5 Attempt ANY FIVE of the following: [10] Classify cholinergic receptors. Name their agonist and antagonist. b) Classify neuromuscular blockers. c) Define cycloplegia and photophobia. d) Enlist the therapeutic uses of anticholinesterases. Name the drugs used treatment of glaucoma. e) What are the contra-indications of phenylephrine as nasal decongestants? O.6 a) Classify adrenergic drugs. Explain the pharmacological actions of adrenaline. Classify \(\beta \) blockers. Discuss the pharmacology, uses and adverse effects of [07] Propranolol. Q.7 a) Explain in detail the mechanism of action, pharmacology and adverse effects of [08]

b) Differentiate between non-depolarizing and polarizing neuromuscular blockers.

Write short notes on ANY THREE of the following:

β agonist as smooth muscle relaxants

Dale's vasomotor reversal.

[07]

[15]

anticholinesterases.

a) Atropineb) Anorectics

c)

Q.8

WINTER - 2014

PURUS - IV (2011 COURSE) : . SUBJECT: PHARAMACEUTICAL CHEMISTRY - VI (ORGANIC)

Day: Tuesday Date: 11-11-2014

Time: 2:00 P. M. To 5:00 P. M. Max Marks. 80

N.B.

- 1) Q. No. 1 and Q. No 5 are **COMPULSORY**. Out of the remaining questions attempt any **TWO** from each section
- 2) Figures to the right indicate FULL marks.
- 3) Both the section should be written in SEPARATE answer books.

SECTION-I

Q.1 Answer any FIVE of the following:

(10)

- a) Unlike benzene pyrrole nitration is performed with acetylnitrate and not with mixture of conc H₂SO₄ and cone HNO₃. Explain.
- b) Write the structure of following heterocyclic
 - i) benzo [g] quinoline
 - ii) furo [3, 2, b] pyridine
 - iii) 3 methoxy 7H azepine
- c) What will happen when pyrrole is treated with C₂H₅O ⁻ N₄ ⁻ (Chloroform and sodium ethoxide)
- d) What will happen when propene is treated with HBr in the presence of peroxide? Justify your answer.
- e) Explain how free radical iodination is difficult to perform.
- f) Pyridine 1 oxide undergoes electrophilic substitutions at C₂ and C₄ position. Explain.
- g) Name medicinally important compounds of indole

Q.2 Answer any THREE of the following:

(15)

- a) Give two methods of synthesis of isoquinoline
- b) Compare and contrast chemical properties of pyrrole and imidazole with respect to electrophilic substitutions.
- c) Describe generation of free radicals.
- d) Discuss autoxidation.
- Q.3 Give methods of synthesis and reactions of any TWO of the following (15) heterocyles.
 - a) Furan
- b) Indole
- c) Thiazole

Q.4		Write short notes on any THREE of the following	(15)	
	a)	Skraup synthesis		
	b)	Friedlander Synthesis		
	c)	Free radical Polymerization		
	d)	Paal Knorr Synthesis		
SECTION II				
Q.5		Answer any FIVE of the following:	(10)	
	a)	What will happen when glucose is treated with 12% HCl		
	b)	What are glycosides?		
	c)	Glucose and fructose produce same osazone when treated with		
		phenylhydrazine. Explain .		
	d)	What are reducing sugars?		
	e)	Why starch gives violet colour when treated with iodine?		
	f)	Why amino acids have high melting points?		
	g)	Give classification of carbohydrates.		
Q.6		Discuss how cyclic structure of glucose is favoured to open chain structure.	(15)	
Q.7	a)	Describe Sanger or Per-Edman methods used to its find sequence of amino	(05)	
		acids in peptides.		
	b)	Give methods of synthesis of amino acids.	(10)	
0.0		Weitershoot and a Company of the Company		
Q.8	2)	Write short notes on any THREE of the following	(15)	
	a)	Mutarotation		
	b)	Structure of sucrose		
	c)	Isoelectric point		
	d)	Protein structure		